A New Iridoid Alkaloid from the Flowers of Plumeria rubra L. cv. acutifolia

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A new iridoid alkaloid containing a spirolactone unit, plumericidine (1), was isolated from the flowers of *Plumeria rubra* L. *cv. Acutifolia*. Its structure was elucidated by spectroscopic evidence and confirmed by X-ray diffraction crystallography. Its anticancer and antiviral activities were evaluated, but found to be insignificant.

Introduction. – The genus *Plumeria* (Apocynaceae family) originated from Central America and consists of about eight species of which many are widely distributed in tropical countries. Phytochemical studies on this genus demonstrated that it is a source of the rather rare lactone-containing iridoids of the plumieride-type [1-4] which possess anticancer activities [5].

P. rubra L. *cv. acutifolia*, one of the two species of the genus occurring in China, is a small tree of which the flowers are used to treat cold and fever, whooping-cough, tracheitis, infective hepatitis, diarrhea, calculus of urethra, and mastitis [6]. Recently, we have isolated two new iridoids from these flowers [7]. In this article, we report on the isolation and structure elucidation of a new alkaloid (*Fig. 1*) from the title plant and on its related bioactivity assay.



Results and Discussion. – *Structure Elucidation.* Plumericidine (1) was obtained as a cubic crystalline compound (MeOH). The molecular formula was established as $C_{13}H_{13}NO_3$ by an M^+ ion peak at m/z 231.0890 (calc. 231.0895) in the HR-EI-MS

¹) Arbitrary atom numbering. For the systematic name, see *Exper. Part.*

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spectrum. In the ¹³C-NMR spectrum, the C-atoms which resonated at δ (C) 149.1, 138.3, and 170.9 could be ascribed to an α,β -unsaturated CO structure. Additionally, an HMBC interaction between H–C(10)¹) with C(12), in addition to the strong absorption band at 1761 cm⁻¹ in the IR spectrum, was also supportive for the presence of an α,β -unsaturated γ -lactone unit in compound **1**. In the ¹H-NMR spectrum, the signal at δ (H) 7.54 for H–C(10) was coupled with a CH H-atom bearing a OH group as indicated by the chemical shift δ (H) 4.47, which also coupled with a Me group (δ (H) 1.37). Further HMBC correlations between Me(14) with C(11), H–C(13) with C(10), and H–C(10) with C(13) indicated that the 1-hydroxyeth-1-yl fragment was the α -side chain of the five-membered α,β -unsaturated lactone.

The downfield signals at $\delta(H)$ 7.43, 8.25, and 8.51 in the ¹H-NMR spectrum of **1** represent an *ABX* coupling system. Together with a strong absorption band (1604 cm⁻¹) in the IR spectrum, it suggested the presence of a pyridine ring in compound **1**. This assumption was verified by the three CH C-atoms ($\delta(C)$ 120.9, 145.0, 150.2) and two quaternary C-atoms at $\delta(C)$ 136.0 and 154.2, observed in the ¹³C-NMR spectrum.

Hitherto, two cyclic fragments of the molecule, the five-membered unsaturated lactone, and the disubstituted pyridine were deduced and left two CH₂ groups to be assigned. On considering the degree of unsaturation, compound **1** should be endowed with a third ring. Thus, the three *multiplets* of H-atom signals at $\delta(H) 3.01-3.15 (2 \text{ H})$, $\delta(H) 2.23-2.32$, and $\delta(H) 2.52-2.60 (1 \text{ H each})$ were assignable to two CH₂ groups of a cyclopentene ring, which was confirmed by the HMBC experiment. HMBC correlations observed between CH₂(7) and C(3), C(4), C(5), C(8), and C(9), and between CH₂(8) and C(3), C(4), C(9), and C(10) established the presence of the cyclopentene moiety in **1**. All the HMBC correlations are listed in the *Table*.

	$\delta(\mathrm{H})$	$\delta(C)$	$H \rightarrow C$ correlations
H-C(2)	8.25 (br. s)	145.0(d)	C(2), C(4), C(5), C(6), C(9)
C(3)		136.0(s)	
C(4)		154.2(s)	
H-C(5)	7.43 $(dd, J = 5.1, 0.9)$	120.9(d)	C(2), C(6), C(7)
H-C(6)	8.51 (d, J = 5.1)	150.2(d)	C(3), C(4), C(5), C(7)
$CH_2(7)$	3.01 - 3.15(m)	29.8(t)	C(3), C(4), C(5), C(6), C(8), C(9)
$CH_2(8)$	2.23 - 2.32(m), 2.52 - 2.60(m)	35.3(t)	C(3), C(4), C(7), C(9), C(10)
C(9)		93.1 (s)	
H - C(10)	7.54 (d, J = 1.2)	149.1(d)	C(8), C(9), C(11), C(12), C(13), C(14)
C(11)		138.3 (s)	
C(12)		170.9 (s)	
H - C(13)	4.47 (dq, J = 6.6, 1.5)	61.6(d)	C(10), C(11), C(12), C(14)
Me(14)	1.37 (d, J = 6.6)	22.4(q)	C(11), C(13)
HO-C(13)	5.38 (d, J = 4.5)		C(11), C(13), C(14)

Table. ¹*H*- and ¹³*C*-*NMR* Data of $\mathbf{1}^1$) (in (D₆)DMSO; δ in ppm, J in Hz)

A single-crystal X-ray diffraction analysis (*Fig.* 2) of plumericidine (1) revealed a unique system with a five-membered ring fused with a pyridine ring, being spiro-linked to an α,β -unsaturated γ -lactone. Intermolecular H-bonds were observed between the HO-C(13) and N(1). On the basis of the configuration of the plumieride-type iridoids

of this genus, the absolute configuration at C(9) and C(13) of the alkaloid was assigned as (S) and (R), respectively, as those of plumieride by biogenetic basis and the literature data [8][9]. To the best of our knowledge, only three alkaloids identified from *Plumeria* genus [10][11], the skeletons of the isolated alkaloids comprise indole, indolizidine, and quinolizidine types. This iridoid alkaloid could contribute to chemotaxonomic significance within this species.



Fig. 2. Single-crystal X-ray structure of 1¹)

Experimental Part

General. All solvents used were of anal. grade (Shanghai Chemical Plant). Column chromatography (CC): silica gel H (SiO₂; 200–300 mesh; Qingdao Marine Chemical Ltd.), Sephadex LH-20 (25–100 µm; Pharmacia). TLC: silica gel GF254 (Yantai Huiyou). M.p.: Leica Galen III apparatus; uncorrected. Optical rotations: Perkin-Elmer 341 polarimeter. IR Spectra: Perkin-Elmer 577 spectrometer with KBr disk. NMR Spectra: Varian INOVA-500 spectrometer operating at 500 and 125 MHz for ¹H and ¹³C, resp.; the detections were carried out at r.t., about 5 mg of samples were dissolved in 0.5 ml DMSO to record the NMR spectra; chemical shifts are given in ppm with TMS as internal reference and coupling constants (J) in Hz. LR- and HR-EI-MS: Finnigan MAT-95 spectrometer.

Plant Material. The flowers of *P. rubra* L. *cv. acutifolia* were collected in March 2007 in Guangdong Province, P. R. China. Authenticated by Prof. *Jin-Gui Shen* of Shanghai Institute of Materia Medica, a voucher specimen (No. 20070310) was deposited in the herbarium of Shanghai Institute of Materia Medica, Shanghai, P. R. China.

Extraction and Isolation. Dried flowers of *P. rubra* L. *cv. acutifolia* (2.9 kg) were extracted with 95% EtOH for three times (1.5 h each). The extracts were concentrated, suspended in H_2O , and then partitioned with petroleum ether, AcOEt, and BuOH sequentially three times each. The AcOEt soluble fraction (2.5 g) was chromatographed on a SiO₂ column eluted with a gradient of CHCl₃/MeOH to afford

three fractions (*Fr.* 1-3). *Fr.* 2 was rechromatographed on SiO₂, and then chromatograhed on *Sephadex LH*-20, yielding compound **1** (20 mg).

Plumericidine (=(7S)-5,6-Dihydro-4'-[(1R)-1-hydroxyethyl]-5'H-spiro[cyclopenta[c]pyridine-7,2'furan]-5'-one; 1). M.p. 214–216°. [α]_D²³ = +17 (c = 0.12, MeOH). IR (KBr): 2989, 1761, 1604, 1425, 1356, 790. ¹H- and ¹³C-NMR data: *Table*. EI-MS: 231. HR-EI-MS: 231.0890 (M^+ , C₁₃H₁₃NO₃⁺; calc. 231.0895).

X-Ray Crystal-Structure Analysis of 1^2). Cubic crystals of 1 were obtained by recrystallization in MeOH. The crystal (0.490 × 0.451 × 0.397 mm) belongs to the orthorhombic system, with the formula C₁₃H₁₃NO₃ (M_w 231.24), space group $P_{2_12_12_1}$ with a = 8.4549(9), b = 8.9753(9), c = 15.6471(16) Å; $a = \beta = \gamma = 90.0^\circ$; V = 1187.4(2) Å³; Z = 4; and $\rho_{calc} = 1.294$ Mg m⁻³. A total of 6183 reflections were collected to a maximum 2θ value of 53.98° by using the Φ/ω scan technique at 293(2) K. The structure was solved by direct methods and was refined by means of the full-matrix least-squares procedure. The collection data were reduced by the Saint program [12] and the empirical absorption correction was performed with the Sadabs program [13]. All non-H-atoms were given anisotropic thermal parameters. The H-atom positions were geometrically idealized and allowed to ride on their parent atoms. The refinement converged to the final R = 0.0423, wR = 0.1069 for 1290 observed reflections ($I > 2\sigma(I)$) and 160 variable parameters.

Cytotoxicity Assay. Cells were cultured in 96-well microtiter plates for the assay. After incubation for 24 h and treatment with 10^{-2} to $10^2 \,\mu\text{M}$ of the test compound for 72 h, growth inhibition of the cancer was investigated. Cells were evaluated by the SRB method (adherent cells: HepG2, KB, and LoVo) or WST-1 method (suspended cell: K562), as described in the literature [14][15]. Adriamycin and taxotere were used as positive controls. However, no activity was detected in these assays (highest concentration tested: 400 μ g ml⁻¹).

Antiviral Assay. HBsAg and HBeAg in the HepG2 2215 cell line supernatant were assayed with commercial enzyme-linked immunosorbent assay (ELISA) kits. The results revealed that **1** inhibited the production of HBsAg (49.6%) at 400 μ g ml⁻¹.

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REFERENCES

- [1] J. J. W. Coppen, A. L. Cobb, *Phytochemistry* **1983**, 22, 125.
- [2] B. S. Siddiqui, A. Naeed, S. Begum, S. Siddiqui, Phytochemistry 1994, 37, 769.
- [3] F. Abe, R.-F. Chen, T. Yamauchi, Chem. Pharm. Bull. 1988, 36, 2784.
- [4] L. B. S. Kardono, S. Tsauri, K. Padmawinata, J. M. Pezzuto, A. D. Kinghorn, J. Nat. Prod. 1990, 53, 1447.
- [5] M. P. Dobhal, G. Li, A. Gryshuk, A. Graham, A. K. Bhatanager, S. D. Khaja, Y. C. Joshi, M. C. Sharma, A. Oseroff, R. K. Pandey, J. Org. Chem. 2004, 69, 6165.
- [6] Editorial Committee of the Administration Bureau of Traditional Chinese Medicine, 'Chinese Materia Medica', Shanghai Science & Technology Press, Shanghai, 1999, Vol. 6, p. 5620.
- [7] G. Ye, Y. L. Yang, G. X. Xia, M. S. Fan, C. G. Huang, Magn. Reson. Chem. 2008, 46, 1195.
- [8] P. J. Stephens, J. J. Pan, F. J. Devlin, K. Krohn, T. Kurtán, J. Org. Chem. 2007, 72, 3521.
- [9] F. Abe, T. Mori, T. Yamauchi, Chem. Pharm. Bull. 1984, 32, 2947.
- [10] O. O. França, R. T. Brown, C. A. M. Santos, Fitoterapia 2000, 71, 208.
- [11] S. N. H. Kazmi, Z. Ahmed, W. Ahmed, A. Malik, Heterocycles 1989, 29, 1901.
- [12] Saint, Program to integrate and reduce raw crystallographic area detector data, Bruker AXS Inc., Madison USA, 1996.

²) CCDC-cd27438 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the *Cambridge Crystallographic Data Centre via* www.ccdc.cam.ac.uk/ data_request/cif.

- [13] G. M. Sheldrick, Sadabs, Program to empirical absorption correction of area detector data, University of Göttingen, Göttingen, 1996.
- [14] P. Skehan, R. Storeng, D. Scudiero, A. Monks, J. McMahon, D. Vistica, J. T. Warren, H. Bokesch, S. Kenney, M. R. Boyd, J. Natl. Cancer Inst. 1990, 82, 1107.
- [15] Y. Zhou, W. Zhu, Q.-X. Zhang, J. Trop. Med. 2005, 5, 580.

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